

# X-ray Crystal Structure of the Nitrogenase Molybdenum–Iron Protein from *Clostridium pasteurianum* at 3.0-Å Resolution<sup>†,‡</sup>

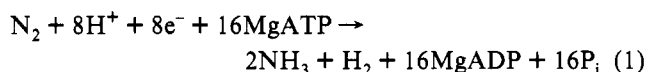
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**ABSTRACT:** The crystal structure of the nitrogenase molybdenum–iron (MoFe) protein from *Clostridium pasteurianum* (Cp1) has been determined at 3.0-Å resolution by a combination of isomorphous replacement, molecular replacement, and noncrystallographic symmetry averaging. The structure of Cp1, including the two types of metal centers associated with the protein (the FeMo-cofactor and the P-cluster pair), is similar to that previously described for the MoFe-protein from *Azotobacter vinelandii* (Av1). Unique features of the Cp1 structure arise from the presence of an ~50-residue insertion in the  $\alpha$  subunit and an ~50-residue deletion in the  $\beta$  subunit. As a consequence, the FeMo-cofactor is more buried in Cp1 than in Av1, since the insertion is located on the surface above the FeMo-cofactor. The location of this insertion near the putative nitrogenase iron protein binding site provides a structural basis for the observation that the nitrogenase proteins from *C. pasteurianum* have low activity with complementary nitrogenase proteins isolated from other organisms. Mechanistic implications of the Cp1 structure for substrate entry/product release, substrate binding to the FeMo-cofactor, and electron- and proton-transfer reactions of nitrogenase are discussed.

Biological nitrogen fixation (reduction of dinitrogen to ammonia) is carried out by a variety of free-living bacteria, cyanobacteria and symbiotic bacteria, in a reaction catalyzed by the nitrogenase enzyme system. The conventional nitrogenase enzyme system consists of two metalloproteins, the molybdenum–iron (MoFe) protein and the iron (Fe) protein [for a review, see Burgess (1984, 1990), Orme-Johnson (1985), Holm and Shimon (1985), Stiefel et al. (1988), Burris (1991), and Smith and Eady (1992)], although homologous alternate nitrogenase systems may be induced under the depletion of molybdenum [for a review, see Bishop et al. (1988) and Eady (1991)]. In addition to the nitrogenase proteins, a source of reducing equivalents (ferredoxin or flavodoxin *in vivo*), MgATP, and protons are required for nitrogen fixation. The overall stoichiometry of biological nitrogen fixation may be represented as (Simpson & Burris, 1984)



The nitrogenase MoFe-protein from *Clostridium pasteurianum* is an  $\alpha_2\beta_2$  tetramer with a molecular mass of ~220 kDa, and the corresponding Fe-protein is a  $\gamma_2$  dimer with a molecular mass of ~60 kDa (Tso, 1974). The MoFe-protein contains two copies of the FeMo-cofactor [M-center; reviewed in Burgess (1990) and Newton (1992)], which are believed to be the substrate binding and reduction sites, and two copies of the P-cluster pair [P-clusters; reviewed in Holm et al. (1990)], which may serve to mediate electron transfer between the Fe-protein and the FeMo-cofactor. The Fe-protein has a single 4Fe:4S cluster which transfers electrons to the MoFe-protein in an ATP-dependent manner. Electron transfer from the Fe-protein to the MoFe-protein involves a cycle of

association and dissociation of the protein complex concomitant with ATP hydrolysis, and the dissociation has been identified as the rate-determining step (Lowe & Thorneley, 1983, 1984; Hageman & Burris, 1978).

The physicochemical properties and the primary structures of Fe- and MoFe-proteins are highly conserved among all nitrogen-fixing bacteria studied so far, but significant differences also exist, especially between the nitrogenase enzyme system of *C. pasteurianum* and that found in other bacteria. The *C. pasteurianum* MoFe-protein (designated Cp1) has the lowest sequence homology with other MoFe-proteins (Ioannidis & Buck, 1987; Wang et al., 1988). The amino acid sequence identity between Cp1 and the *Azotobacter vinelandii* MoFe-protein (designated Av1) is ~36%. Additionally, Cp1 has a long insertion (~50 amino acid residues) in the  $\alpha$  subunit and a long deletion (~50 amino acid residues) in the  $\beta$  subunit (Wang et al., 1988). As a result of these sequence changes, the nitrogenase components from *C. pasteurianum* have a distinctly low capacity to form an active hybrid enzyme with the complementary components of other organisms (Emerich & Burris, 1978). Differences between Cp1 and Av1 in both the relative reduction sequence of the metal centers and their measured midpoint potentials have been observed when redox titrations were monitored by electron paramagnetic resonance (EPR) (Morgan et al., 1987). Since similar studies of the isolated cofactors from each species have shown them to be essentially identical (Morgan et al., 1987), the difference in the behavior of these two proteins probably results from the constraints imposed by each protein on an identical cofactor. The *C. pasteurianum* nitrogenase enzyme system is also less sensitive to  $\text{H}_2$  as an inhibitor (Guth & Burris, 1983) and shows a higher specificity for nucleotides (Weston et al., 1983).

The crystal structure of nitrogenase MoFe-protein from *A. vinelandii*, including the structure of the FeMo-cofactor and the P-cluster pair, has been reported in Kim and Rees (1992a,b). In this paper, the tertiary and quaternary structures of the nitrogenase MoFe-protein from *C. pasteurianum* are

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<sup>‡</sup> The crystallographic coordinates have been deposited in the Brookhaven Protein Data Bank under the file name 1MIO.

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presented, based on a 3.0-Å-resolution X-ray crystallographic analysis. The structural comparison of Av1 versus Cp1 and the functional implications of the nitrogenase MoFe-protein in dinitrogen reduction, including substrate entry/product release, proton-transfer pathways, electron-transfer pathways, and Fe-protein binding sites, are also discussed.

### Structure Determination

The structure of Cp1 was solved by a combination of single isomorphous replacement (SIR), molecular replacement, and noncrystallographic symmetry (NCS) averaging both within and between two crystal forms (Rossmann, 1972; Bricogne, 1976). The purification, crystallization, and heavy atom derivative screening of nitrogenase MoFe-proteins have been previously described (Kim & Rees, 1992a). Two crystal forms of the MoFe-protein in space group  $P2_1$  were prepared from *C. pasteurianum*, designated Mg1 and Cs1, with unit cell constants  $a = 70.0$  Å,  $b = 151.3$  Å,  $c = 121.9$  Å, and  $\beta = 110.4^\circ$  and  $a = 87.9$  Å,  $b = 171.4$  Å,  $c = 73.6$  Å, and  $\beta = 91.5^\circ$ , respectively. Both crystal forms contain one tetramer molecule in an asymmetric unit. The Mg1 crystal form is similar to a form previously described (Weininger & Mortenson, 1982). Each crystal form was derivatized using (ethylmercuri)thiosalicylate (EMTS), and four common EMTS binding sites which are related by the 2-fold NCS axis were found for both crystal forms (Table I). The NCS relationships both within and between crystal forms (Av1, Mg1, and Cs1) were determined from rotation functions (Rossmann & Blow, 1962; Crowther, 1972) and translation functions (Crowther & Blow, 1967) and were confirmed by the heavy atom locations and the FeMo-cofactor and P-cluster pair locations. Model phases from the Av1 structure oriented in the Cp1 unit cell were combined with the SIR phases of the Mg1 and Cs1 crystal forms, and the combined phases were subsequently refined by averaging both within and between the two different crystal forms (Bricogne, 1976) using the 2-fold NCS in both crystal forms. The final  $R$  factor of the 4-fold averaging was  $\sim 22\%$  and that of the subsequent 2-fold averaging within the Mg1 crystal form was  $\sim 18\%$ . The Mg1 crystal form was used for further crystallographic analysis; an atomic model of the Cs1 crystal form has not yet been built and refined. The averaged Mg1 electron density map was of sufficient quality to trace 1820 of the 1980 amino acid residues. The long inserted polypeptide in the  $\alpha$  subunit ( $\alpha 375$ – $\alpha 430$ ) did not appear in the averaged map because the Av1 envelope was used for averaging. Hence this region was built following inspection of  $2F_o - F_c$  and  $F_o - F_c$  maps. The initial model was built into the averaged electron density map using the graphics program TOM/FRODO (Jones, 1985) implemented with the fragment fitting option (Jones & Thirup, 1986) and subsequently refined using the restrained least squares program TNT (Tronrud et al., 1987). The resultant  $2F_o - F_c$  and  $F_o - F_c$  maps and the electron density maps obtained by iterative cycles of model building, refinement, phase combination, and NCS averaging were used for further model building. The simulated annealing protocol in the X-PLOR (Brünger, 1988) program was used during the final stages of coordinate refinement.

The current model contains 1964 amino acid residues (of 1980 total residues) with 15 273 non-hydrogen atoms (99% complete). This model has presently been refined to a crystallographic  $R$  factor of 0.18 (10–3.0 Å) with root mean square deviations from ideal bond distances and angles of 0.018 Å and  $3.9^\circ$ , respectively (Table I). The correctness of the chain trace is further supported by the location of the

Table I: Structure Determination and Refinement Statistics for the *C. pasteurianum* MoFe-Protein (Cp1)<sup>a</sup>

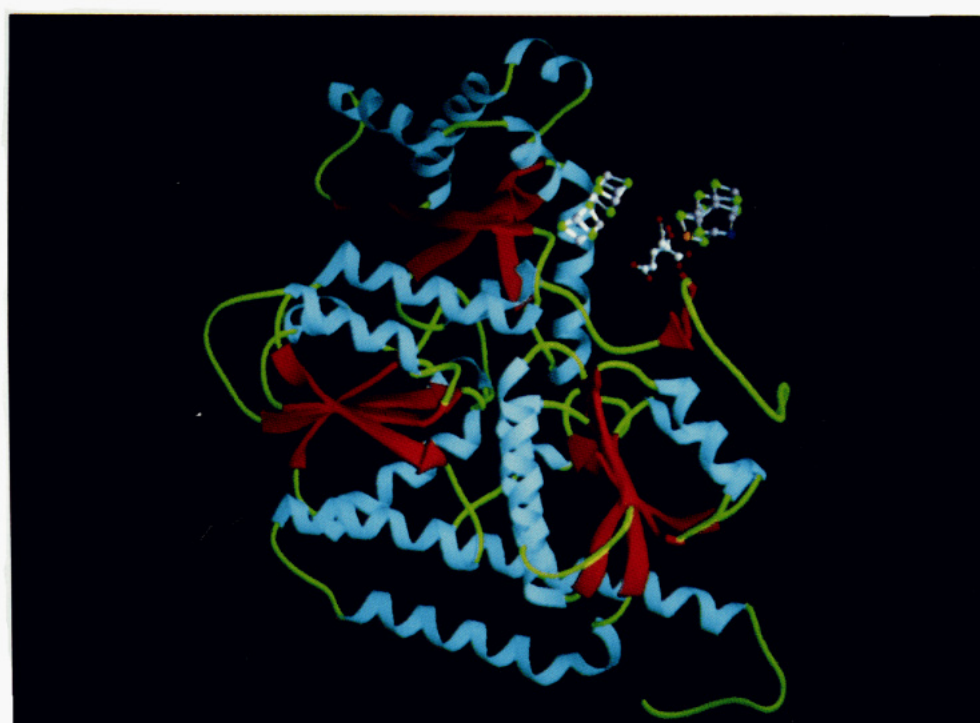
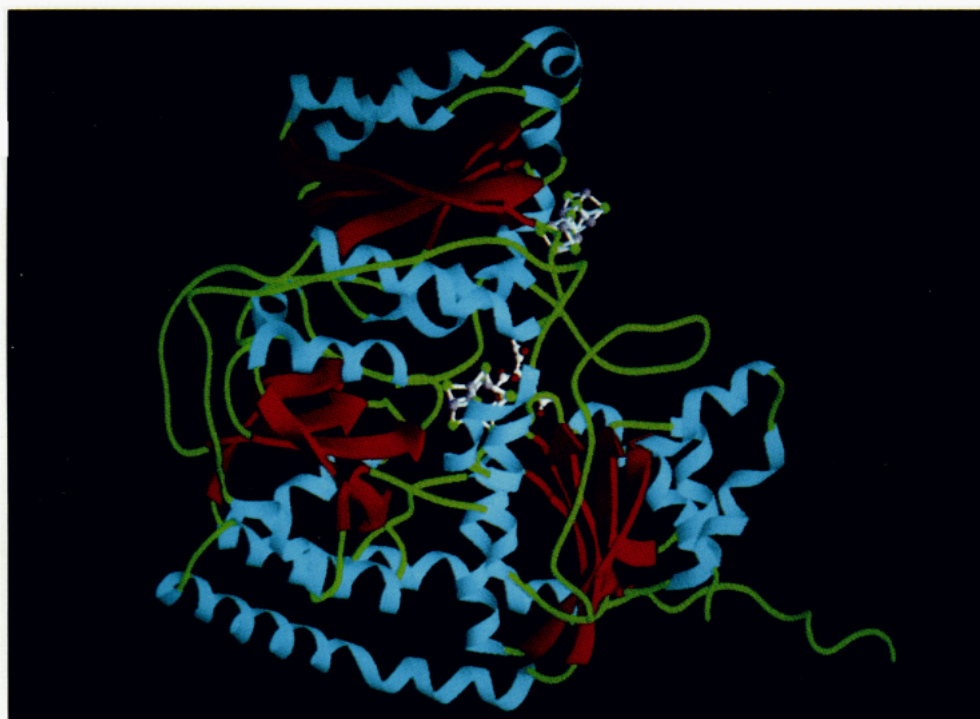
(A) Heavy Atom Binding Sites					
derivative	x	y	z	rel occupancy	location
EMTS (Mg1)	0.356	0.073	0.858	0.23	Cys $\alpha$ 302
	0.125	0.295	0.105	0.22	Cys $\alpha$ 302
					(NCS related)
	0.579	-0.264	0.451	0.15	Cys $\beta$ 257
	-0.348	-0.161	0.104	0.15	Cys $\beta$ 257
					(NCS related)
EMTS (Cs1)	0.416	0.231	-0.105	0.17	b
	-0.033	-0.254	0.071	0.16	
	0.737	0.021	0.719	0.14	
	0.019	-0.008	0.924	0.11	
(B) Model Status (Mg1)					
parameter				value	
no. of amino acid residues				1964/1980	
total non-hydrogen atoms				15273 (99% complete)	
no. of water molecules				0	
missing residues				$\alpha$ 527- $\alpha$ 534	
no. of Ramachandran outliers				18/1960	
(C) Refinement Statistics					
				before refinement	after refinement
R factor (10.0-3.0 Å)				0.39	0.18
rms deviation of					
bond length (Å)				0.024	0.018
bond angles (deg)				2.90	3.9
dihedral angles (deg)				28.3	25.3
improper torsion (deg)				1.66	1.50

<sup>a</sup> The Mg1 native data set collected by a Siemens multiwire area detector at room temperature contains 38 527 reflections and is 78% complete to 3.0-Å resolution with a merging  $R$  factor of 0.07. The Cs1 native data set collected by a Siemens area detector at room temperature contains 33 013 reflections and is 74% complete to 3.0-Å resolution with a merging  $R$  factor of 0.084. Heavy atom derivatives were prepared with (ethylmercuri)thiosalicylate (EMTS) for both the Mg1 and Cs1 crystal forms, and the derivative data sets were also collected by an area detector (Kim & Rees, 1992a). Locations of the heavy atoms were determined from difference Patterson maps with the aid of molecular replacement phases and refined with the program HEAVY (Terwilliger et al., 1987). The mean figure of merit and phasing power of both the Mg1 and Cs1 heavy atom derivatives (EMTS) are about 0.36 and 1.37, respectively. TOM/FRODO (Jones, 1985) was used for model building, and X-PLOR (Brünger, 1988) was used in the final refinements with PARAM19X.PRO as the parameter file. <sup>b</sup> The binding locations for the Cs1 heavy atom sites presumably correspond to the Mg1 sites, but this has not been directly established.

heavy atom binding sites (Table I) in addition to the similar folding of both the  $\alpha$  and  $\beta$  subunits. The EMTS sites are found coordinated to nonconserved Cys residues ( $\alpha 302$  and  $\beta 257$ ). The model has also been examined using the 3D–1D profile method (Lüthy et al., 1992), and all the residues have reasonable average 3D–1D scores except for the region  $\alpha 380$ – $\alpha 391$  in which electron density is diffuse. Residue numbers are prefixed with either  $\alpha$  or  $\beta$  to indicate the appropriate subunit, except where interactions in the tetramer are described, in which case the prefixes A and C designate the two distinct  $\alpha$  subunits while the prefixes B and D designate the two distinct  $\beta$  subunits present in the tetramer.

### Description of the Protein Structure

The  $\alpha$  subunit of Cp1 consists of 533 amino acid residues, which is the longest among all known MoFe-proteins (Wang et al., 1988). The three-dimensional structure of the  $\alpha$  subunit is similar to that of Av1, with the exception of a long inserted loop region ( $\alpha 375$ – $\alpha 430$ ). The  $\alpha$  subunit of Cp1 consists of three domains of the  $\alpha/\beta$  type with some extra helices (Figure 1a,b). The overall shape of the  $\alpha$  subunit may be described as a cloverleaf. Domain I is composed of seven helices, four



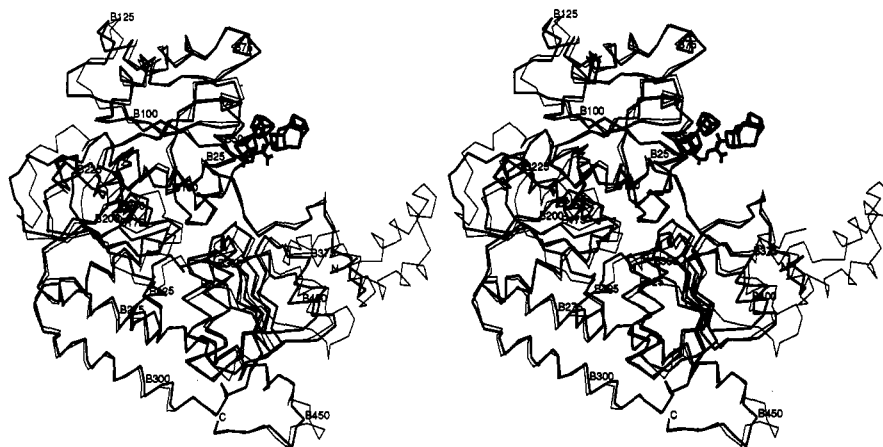


FIGURE 1: (a, left page, top) RIBBONS (Carson & Bugg, 1986) diagram of the polypeptide fold of the  $\alpha$  subunit of Cp1.  $\alpha$ -Helical regions are in light blue,  $\beta$ -sheet regions are in red, and the remaining regions of the polypeptide chain are in green. The three domains are designated I, II, and III, respectively, counterclockwise from the top. The FeMo-cofactor is located in the center of the figure, and the P-cluster pair is on the top right side. (b, left page, middle)  $\alpha$  chain trace of the  $\alpha$  subunit of Cp1 superimposed on that of Av1. The rms deviation between common  $\alpha$  positions is 1.57 Å. The  $\alpha$  subunits of Cp1 and Av1 are represented by the thick and thin lines, respectively. The view is approximately from the same orientation as that in (a). (c, left page, bottom) RIBBONS diagram of the polypeptide fold of the  $\beta$  subunit of Cp1.  $\alpha$ -Helical regions are in light blue,  $\beta$ -sheet regions are in red, and the remaining regions of the polypeptide chain are in green. The three domains are designated I', II', and III', respectively, counterclockwise from the top. The left-most metal center is the P-cluster pair, while the FeMo-cofactor is to the right of the P-cluster pair. (d, above)  $\alpha$  chain trace of the  $\beta$  subunit of Cp1 superimposed on that of Av1. The rms deviation between common  $\alpha$  positions is 1.61 Å. The  $\beta$  subunits of Cp1 and Av1 are represented by the thick and thin lines, respectively. The view is approximately from the same orientation as that in (c).

parallel  $\beta$  strands, and one antiparallel  $\beta$  strand which is provided by the  $\beta$  subunit; domain II is composed of six helices, four parallel  $\beta$  strands, and one antiparallel  $\beta$  strand; and domain III is composed of ten helices, five parallel  $\beta$  strands, and one antiparallel  $\beta$  strand which is from the N-terminus of the  $\alpha$  subunit. There is a wide and shallow cleft between the three domains, and the FeMo-cofactor sits beneath the bottom of this cleft. This cleft has been proposed as a Fe-protein subunit binding site in Av1 (Kim & Rees, 1992b). Residues  $\alpha 375$ – $\alpha 430$ , which are absent in Av1, are located between domains II and III; in particular,  $\alpha 383$ – $\alpha 397$  are located above the cleft providing an additional polypeptide environment in the vicinity of the FeMo-cofactor.

The  $\beta$  subunit of Cp1 consists of 457 amino acid residues, which is the shortest among all known MoFe-proteins (Wang et al., 1988). The structure of the  $\beta$  subunit is also similar to that of Av1 except for the N-terminus ( $\sim 10$  residues of Cp1) and loop regions at the surface where deletions or insertions occur (Figure 1c,d). The overall folding of the  $\beta$  subunit, which is composed of three  $\alpha/\beta$ -type domains, is similar to that of the  $\alpha$  subunit, as has been observed in the Av1 structure. However, significant differences also exist, in particular at the N-terminus, in the  $\alpha 375$ – $\alpha 430$  region, and in surface loop regions. In the  $\beta$  subunit, domain I' is made up of seven helices and four parallel  $\beta$  strands; domain II' is made up of six helices and four parallel  $\beta$  strands; and domain III' is made up of six helices and five parallel  $\beta$  strands. Analogous to the  $\alpha$  subunit, there is a cleft between the three domains, which has been proposed to provide part of the Fe-protein subunit binding site in Av1. The C-terminus of the  $\beta$  subunit, which is located at the interface of the B and D subunits, is not exposed to solvent and is very well ordered in both Cp1 and Av1, while the C-terminus of the  $\alpha$  subunit is exposed to solvent and is not as well ordered in Cp1.

The  $\alpha\beta$  subunit pair consists of six  $\alpha/\beta$ -type domains which are arranged like a six-petaled flower, with the P-cluster pair located in the center of the dimer like a pistil (Figure 2). The  $\alpha$  subunit and the  $\beta$  subunit of the MoFe-protein are related by an approximate 2-fold axis which passes through the center

of the P-cluster pair, and there are two wide and shallow clefts around the P-cluster pair which may provide the binding site for the dimeric Fe-protein. The  $\alpha$  and  $\beta$  subunits are in very close contact with each other. Domain I of the  $\alpha$  subunit contacts domains I' and III' of the  $\beta$  subunit, and domain I' of the  $\beta$  subunit contacts domains I and III of the  $\alpha$  subunit. Domain I of the  $\alpha$  subunit and domain I' of the  $\beta$  subunit are bridged by the P-cluster pair. An antiparallel  $\beta$ -sheet arrangement between  $\beta 15$ – $\beta 18$  and  $\alpha 105$ – $\alpha 109$  also contributes to the  $\alpha\beta$  subunit interface. In addition to these general contacts, salt bridge, hydrophobic, and hydrogen-bonding interactions between the two subunits are also important for dimerization, as is observed in Av1.

As mentioned above, the overall structure of Cp1 is similar to that of Av1. In particular, the tertiary structure around the metal centers is well conserved between the two MoFe-proteins. However, significant differences also exist. The overall dimensions of the  $\alpha_2\beta_2$  Cp1 tetramer are  $\sim 70$  Å  $\times$  80 Å  $\times$  120 Å, and its general shape is slightly elongated along one axis compared to that of Av1 due to the  $\sim 50$ -residue insertion in the  $\alpha$  subunit (Figure 3). The two  $\alpha\beta$  subunit pairs are related by the 2-fold NCS rotation axis used for initial phase refinement. Even though the  $\alpha$  and  $\beta$  subunits in an  $\alpha\beta$  subunit pair are also related by an approximate 2-fold rotation axis, the MoFe-protein tetramer does not exhibit 222 symmetry, which had been proposed by a low-resolution rotation function study (Yamane et al., 1982), because the  $\alpha\beta$  2-fold and the tetramer 2-fold axes are not perpendicular and do not intersect each other. Consequently, the organization of  $\alpha$  and  $\beta$  subunits in the MoFe-protein tetramer differs from that of hemoglobin (Perutz et al., 1960), where the homologous subunits are arranged with approximate 222 symmetry.

The tetramer interface is generated by extensive interactions between domains II' and III' of the two  $\beta$  subunits, along with some additional interactions involving domain III of each  $\alpha$  subunit. Packing between helices from the  $\beta$  subunit (B181–B194, B290–B310, B270–B284, B427–B447, and the corresponding regions of subunit D) dominates the interactions at the tetramer interface, along with additional contributions



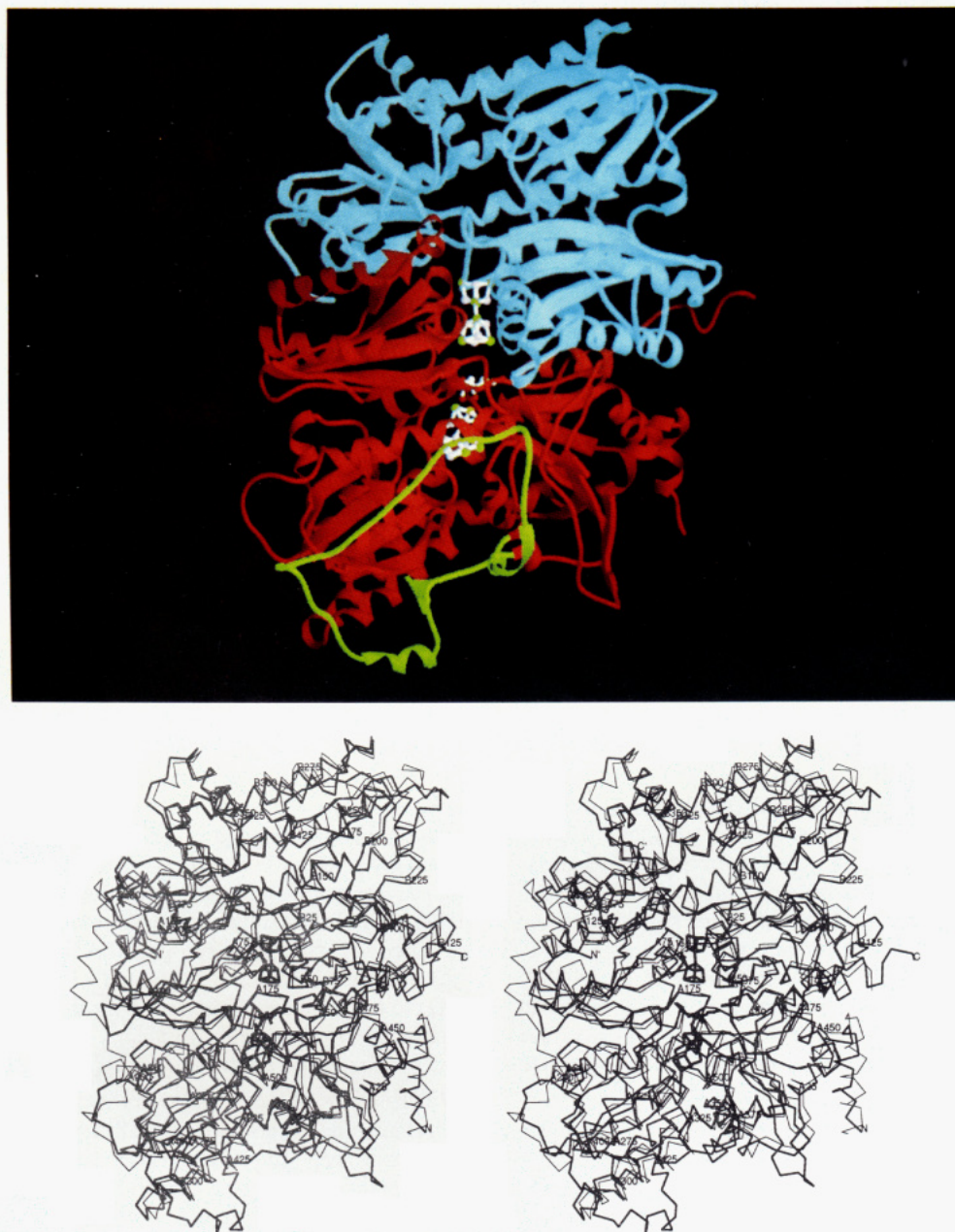


FIGURE 2: (a, top) RIBBONS diagram of the polypeptide fold of an  $\alpha\beta$  subunit pair. The  $\alpha$  subunit is in red and the  $\beta$  subunit is in light blue. The long inserted loop in the  $\alpha$  subunit, which is unique to Cp1, is in green. The view is down the 2-fold axis through the P-cluster pair that approximately relates the  $\alpha$  and  $\beta$  subunits. The P-cluster pair is located in the center of the figure, and the FeMo-cofactor is located below the P-cluster pair. (b, bottom) Ca chain trace of the  $\alpha\beta$  subunit pair of Cp1 superimposed on that of Av1. The rms deviation between 880 common Ca positions is 1.63 Å. The  $\alpha\beta$  subunit pairs of Cp1 and Av1 are represented by the thick and thin lines, respectively. The view is approximately from the same orientation as that in (a).

from helices in the  $\alpha$  subunit (A289–A302, A306–A334, A493–A507, and the corresponding regions of subunit C). The helical interactions appear to provide a major driving force for tetramerization. As a result of the six-helix barrel in the center of the tetramer, the MoFe-protein has a large channel  $\sim 8$ – $10$  Å wide and  $\sim 35$  Å in length. In addition to these helical packing interactions, electrostatic, hydrophobic, and hydrogen-bonding interactions are important for tetramerization, as is observed in Av1.

The divalent cation sites found in the Av1 structure (Kim & Rees, 1992b) also exist in Cp1, and the tertiary structure around these sites is well conserved. On the basis of the electron density value, temperature factor, and the coordination environment, these sites could be assigned as  $\text{Ca}^{2+}$  (or, less likely,  $\text{Mg}^{2+}$ ). This ion has an octahedral coordination environment provided by the carboxyl oxygens of Glu B62, Asp D301, Asp D305, the carbonyl oxygen of Lys B61, and

probably two water molecules. Gln A472, Lys A473, Thr B212, Phe A469, Phe B60, Tyr B420, and Tyr B421 are located outside the immediate coordination sphere. The site is buried at the interface of two  $\alpha\beta$  dimers and is  $\sim 25$  and  $\sim 21$  Å away from the P-cluster pair and FeMo-cofactor, respectively. This site does not appear to have a functional role, but rather may serve to stabilize the subunit associations in the tetramer. Neuraminidase (Varghese et al., 1983) and some spherical plant viruses (Hogle et al., 1983) also have divalent cations ( $\text{Ca}^{2+}$ ) at subunit interfaces that function in the stabilization of subunit contacts.

No disulfide bonds are present in either the Cp1 or Av1 structures, and no ATP/ADP is found in Cp1 and Av1, although some experimental studies have indicated that oxidized MoFe-protein can bind MgADP (Miller & Eady, 1989).





FIGURE 3: RIBBONS diagram of the polypeptide fold of the  $\alpha_2\beta_2$  MoFe-protein tetramer. The view is down the tetramer 2-fold axis.  $\alpha$ -Helical regions are in light blue,  $\beta$ -sheet regions are in red, and the remaining regions of the polypeptide chain are in green.

#### Structures of the Metal Centers

The structures of the metal centers in Av1 have been described in Kim and Rees (1992a) and verified by high-resolution X-ray diffraction analysis (Chan et al., 1993). The FeMo-cofactor and P-cluster pair structures of Cp1 appear identical to those of Av1. The FeMo-cofactor contains 4Fe:3S and 1Mo:3Fe:3S clusters that are bridged by three nonprotein ligands (Figure 4a). Two of the bridging ligands are assigned as sulfurs, while the chemical identity of the third ligand is still ambiguous and could be either a well-ordered O/N species or a less well ordered S species in Av1. The electron density of this site looks more like a sulfur in Cp1. The Mo site has the highest electron density value in the 3.0-Å-resolution  $2F_o - F_c$  map. Although the other atomic positions are not resolved, the FeMo-cofactor model generally matches the electron density well. Homocitrate, an essential component of FeMo-cofactor (Hoover et al., 1989), is coordinated through hydroxyl and carboxyl oxygens to the Mo site. The importance of the hydroxyl group in homocitrate had been implicated by studies in which the homocitrate was replaced with other carboxylic acids (Hoover et al., 1988). The FeMo-cofactor is attached to the  $\alpha$  subunit through two protein ligands, Cys  $\alpha 262$ , which coordinates Fe1, and His  $\alpha 482$ , which coordinates Mo along with homocitrate.

A ball and stick model of the P-cluster pair is presented in Figure 4b. The P-cluster pair consists of two 4Fe:4S clusters, as had been suggested by Mössbauer and extrusion studies (Huynh et al., 1980; Kurtz et al., 1979), that are bridged by two cysteine thiol ligands. X-PLOR refinement results indicate that the two 4Fe:4S clusters are additionally linked by a disulfide bond (S-S bond distance =  $\sim 2.2$ – $2.3$  Å) formed between the sulfurs from each 4Fe:4S cluster. The existence

of a disulfide bond in the P-cluster pair is consistent with the Av1 P-cluster pair structure (Chan et al., 1993). The P-cluster pair is attached between the  $\alpha$  and  $\beta$  subunits through seven protein ligands: Cys  $\alpha 53$ , Cys  $\alpha 79$ , Cys  $\alpha 145$ , Cys  $\beta 23$ , Cys  $\beta 48$ , Cys  $\beta 106$ , and Ser  $\beta 141$ . Cys  $\alpha 79$  bridges Fe4 and Fe5 and Cys  $\beta 48$  bridges Fe1 and Fe8. Cys  $\alpha 53$ , Cys  $\alpha 145$ , Cys  $\beta 23$ , and Cys  $\beta 106$  coordinate the remaining Fe sites. Ser  $\beta 141$  is close to Fe6 and may coordinate this site along with Cys  $\beta 106$ .

#### Environment of the FeMo-Cofactor

The FeMo-cofactor, which almost certainly provides the substrate binding and reduction site, can be isolated intact from the MoFe-protein (Shah & Brill, 1977). In its isolated form, however, it no longer catalyzes dinitrogen reduction (Shah & Brill, 1977). Because the FeMo-cofactor must be protein-bound in order to reduce substrate, the polypeptide environment must contribute to its substrate binding and reduction properties. The FeMo-cofactor and its surrounding residues are represented in Figure 5a. The FeMo-cofactor is buried at least 10 Å from the protein surface, and the polypeptide environment around the FeMo-cofactor is primarily provided by the  $\alpha$  subunit. The FeMo-cofactor is buried more deeply in Cp1 than in Av1 because the  $\alpha 383$ – $\alpha 397$  loop, which is unique to Cp1, is located above the FeMo-cofactor and partially occupies the cleft as shown in Figures 5c and 7c. Cys  $\alpha 262$  and His  $\alpha 482$  are coordinated to the FeMo-cofactor and Ser  $\alpha 265$  is hydrogen bonded to the  $S_\gamma$  of Cys  $\alpha 262$ . These residues are strictly conserved among all known MoFe-protein sequences and are structurally important, along with conserved Gly residues (Gly  $\alpha 344$  and Gly  $\alpha 345$ ) whose amide



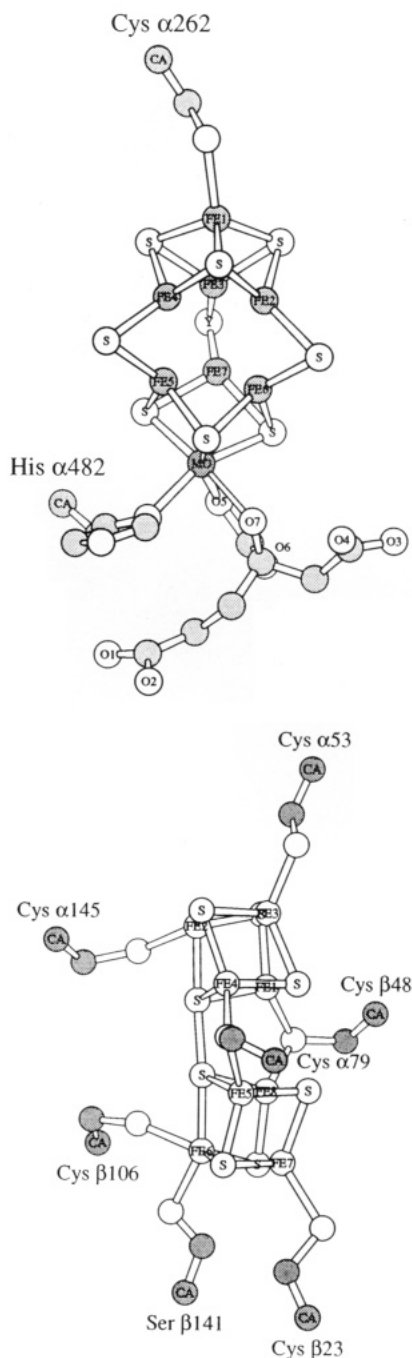


FIGURE 4: (a, top) Ball and stick model of the FeMo-cofactor with protein ligands, drawn with the MOLSCRIPT program (Kraulis, 1991). The Y ligand could be S or a well-ordered O/N species (less likely). (b, bottom) Ball and stick model of the P-cluster pair with protein ligands, drawn with the MOLSCRIPT program.

NH groups are hydrogen bonded to sulfurs in the FeMo-cofactor. Other highly conserved residues near the FeMo-cofactor include Arg  $\alpha$ 87 and Arg  $\alpha$ 347, which can potentially form hydrogen bonds to sulfurs in the FeMo-cofactor and may serve to electrostatically stabilize the FeMo-cofactor and/or partially reduced intermediate formed during substrate reduction; His  $\alpha$ 186, which is hydrogen bonded to a bridging sulfur and may participate in proton-transfer reactions; Gln  $\alpha$ 182, Glu  $\alpha$ 368, Glu  $\alpha$ 467, Gln  $\alpha$ 480, and His  $\alpha$ 482, which are near the homocitrate and interact with this group either directly or through water molecules; and aromatic and hydrophobic residues, such as Tyr  $\alpha$ 216, Phe  $\alpha$ 369, Val  $\alpha$ 61, and Ile  $\alpha$ 220.

### Substrate Entry and Product Release

Even though the FeMo-cofactor is buried at least 10 Å below the protein surface and there are no permanent channels between the protein surface and the FeMo-cofactor (Kim & Rees, 1992b), there are two clefts which could be potentially utilized for substrate entry/product release and/or  $\text{H}_3\text{O}^+$  transfer to the active site. The first cleft, which is made up of five stretches of polypeptide chain ( $\alpha$ 255– $\alpha$ 275,  $\alpha$ 279– $\alpha$ 301,  $\alpha$ 338– $\alpha$ 359,  $\alpha$ 363– $\alpha$ 378, and  $\alpha$ 422– $\alpha$ 436), exists between domains II and III and has a funnel shape with an outer diameter of  $\sim$ 4–10 Å (Figure 5b). The second cleft, also made up of five stretches of polypeptide chain ( $\alpha$ 172– $\alpha$ 199,  $\alpha$ 341– $\alpha$ 352,  $\alpha$ 365– $\alpha$ 400,  $\alpha$ 256– $\alpha$ 276, and  $\alpha$ 34– $\alpha$ 43), exists between the three domains of the  $\alpha$  subunit and has a funnel shape with an outer diameter of  $\sim$ 3–10 Å (Figure 5c). These two clefts are near the one of the putative Fe-protein subunit binding sites (see below) and may also provide possible openings for cofactor insertion during biosynthesis of the MoFe-protein. However, neither of these two clefts is wide enough to allow free diffusion of either substrate/product or  $\text{H}_3\text{O}^+$ . Therefore, it seems reasonable to assume that there must be structural fluctuations or conformational changes to allow the diffusion of ligands into and away from the FeMo-cofactor, analogous to the binding and release of oxygen to the buried hemes in globins (Case & Karplus, 1979).

In addition to the physiological reactions of  $\text{N}_2$  reduction and  $\text{H}_2$  formation, nitrogenase catalyzes a wide variety of reactions involving small unsaturated molecules, such as azides, nitrous oxide, nitriles, isonitriles, and alkynes (Burgess, 1985). Functionally, the details of the interaction between various substrates and the FeMo-cofactor are central to understanding the catalytic properties of nitrogenase. While many modes of substrate binding to one or more of the Fe, Mo, and/or S sites are possible [see, for example, Deng and Hoffmann (1993)], the six trigonally coordinated iron sites of the FeMo-cofactor are particularly intriguing for substrate binding. On the basis of the FeMo-cofactor structure, three general types of binding modes of substrates to these coordinatively unsaturated irons may be envisioned: (i) Substrates could bind in an end-on fashion to one of the six central iron sites which are bridged by nonprotein ligands. It is also conceivable that some substrates could displace the bridging ligand Y, if this ligand is not a sulfur, and interact simultaneously with the two adjacent iron sites. (ii) Some small substrates, such as  $\text{N}_2$  and/or  $\text{H}^+$ , could occupy the central cavity in the FeMo-cofactor, thereby replacing the weak iron–iron bonds with Fe–substrate bonds (Chan et al., 1993). (iii) Substrates could bind to the exterior surface of the FeMo-cofactor. Three sets of cyclic eight-membered rings occur on the exterior surface of the FeMo-cofactor, consisting of alternating S(Y) and Fe sites. These arrangements may create the equivalent of a small region of an iron surface, each containing four iron atoms, and substrates could bind to those surfaces so as to simultaneously interact with up to four irons. The presence of multiple, potential substrate binding sites in the FeMo-cofactor may be related to the noncompetitive kinetics observed between  $\text{N}_2$  and other substrates (with the exception of  $\text{N}_2\text{O}$ ) (Rivera-Ortiz & Burris, 1975).

### Proton Transfer

In order to reduce dinitrogen to ammonia, a supply of protons is essential in addition to dinitrogen. In the vicinity of the

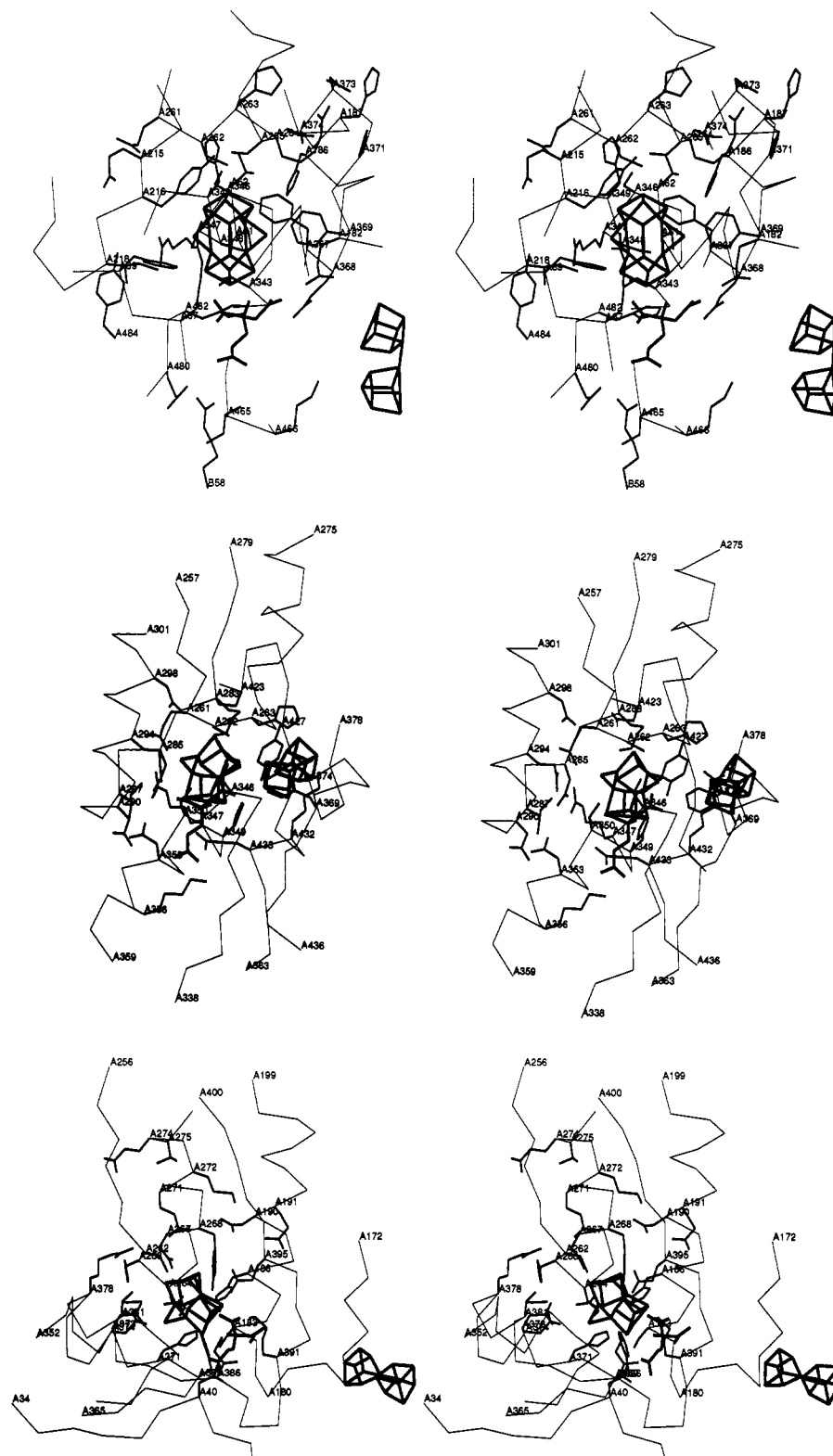


FIGURE 5: (a, top) Stereoview of the polypeptide environment surrounding the FeMo-cofactor. (b, middle) Stereoview of a cleft around the FeMo-cofactor that could be utilized for substrate entry/product release and/or  $\text{H}_3\text{O}^+$  transport. (c, bottom) Stereoview of the other cleft around the FeMo-cofactor which could be utilized for substrate entry/product release and/or  $\text{H}_3\text{O}^+$  transport.

FeMo-cofactor, several potential pathways occur which could transfer protons to the FeMo-cofactor: Glu  $\alpha$ 368 and Glu  $\alpha$ 467 to homocitrate to FeMo-cofactor, and His  $\alpha$ 186 to FeMo-cofactor. If an intermediate that is a sufficiently strong base is generated during dinitrogen reduction, it is possible that protons may also be transferred from Arg  $\alpha$ 347 and/or Arg  $\alpha$ 87 to the FeMo-cofactor. The two potential channels around the FeMo-cofactor shown in Figure 5b,c could be utilized for

the proton transfer to the vicinity of the cofactor. Charged or hydrophilic residues lining these potential channels, such as Lys  $\alpha$ 432, Lys  $\alpha$ 283, His  $\alpha$ 263, His  $\alpha$ 186, His  $\alpha$ 349, Glu  $\alpha$ 433, Arg  $\alpha$ 378, His  $\alpha$ 187, Glu  $\alpha$ 294, Asn  $\alpha$ 285, Glu  $\alpha$ 271, Asp  $\alpha$ 298, His  $\alpha$ 371, Asn  $\alpha$ 267, Arg  $\alpha$ 264, Asp  $\alpha$ 373, and Asp  $\alpha$ 374, may facilitate proton transfer during the substrate reduction. Two patches of histidines at the entrance to these clefts might represent the initial site for proton binding; these



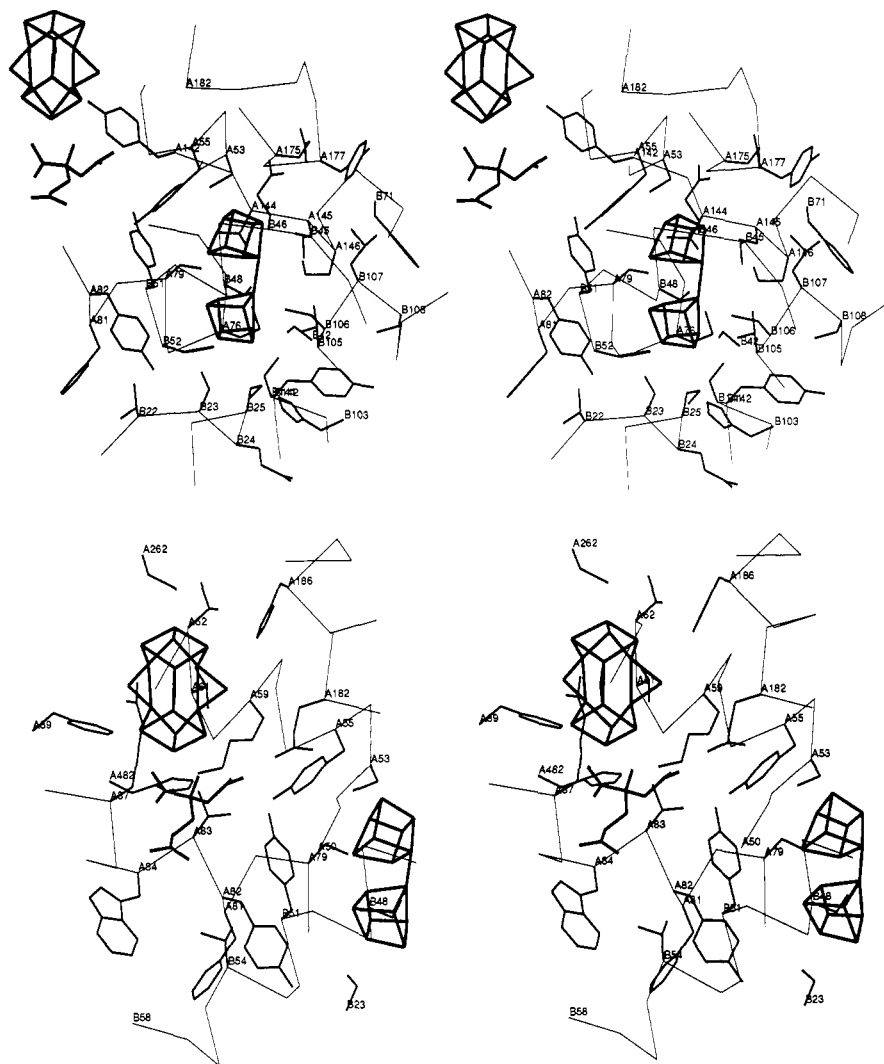


FIGURE 6: (a, top) Stereoview of the polypeptide environment surrounding the P-cluster pair. (b, bottom) Stereoview of the polypeptide environment between the FeMo-cofactor and the P-cluster pair.

patches contain the imidazole side chains of residues  $\alpha 187$ ,  $\alpha 371$ , and  $\alpha 263$ ; and of residues  $\alpha 349$  and  $\alpha 435$ . The presence of multiple, potential proton-transfer routes suggests that there is not a unique pathway by which protons are shuttled from the surface to the active site. The proton-transfer pathway need not be unique, although there may be a major pathway, considering that at least eight protons are required for reduction of one dinitrogen (Simpson & Burris, 1984).

#### Environment of the P-Cluster Pair

The P-cluster pair, which may function in electron transfer between the 4Fe:4S cluster of Fe-protein and the FeMo-cofactor, is composed of two 4Fe:4S clusters bridged by two cysteine thiol ligands and one disulfide bond between two cluster sulfurs. In spite of the functional importance and novel structural features of the P-cluster pair, nitrogenase research has not generally focused on the P-cluster pair. Consequently, the understanding of how the P-cluster pair is synthesized and incorporated in MoFe-protein is quite limited, compared to that of the FeMo-cofactor. The P-cluster pair and its surrounding residues are shown in Figure 6a. The P-cluster pair bridges the  $\alpha$  and  $\beta$  subunits and is completely buried at least 12 Å from the protein surface. The highly buried nature of the P-cluster pair and its polypeptide environment may

provide an explanation for why the MoFe-protein requires a special electron donor, i.e., the Fe-protein. The polypeptide environment around the P-cluster pair is mainly provided by aromatic and hydrophobic residues, including Tyr  $\alpha 55$ , Pro  $\alpha 76$ , Tyr  $\alpha 82$ , Pro  $\alpha 146$ , Pro  $\beta 25$ , Tyr  $\alpha 142$ , Tyr  $\beta 51$ , Tyr  $\beta 142$ , and Phe  $\beta 71$ . These residues are highly conserved among all known MoFe-protein sequences. Cys  $\alpha 53$ , Cys  $\alpha 79$ , Cys  $\alpha 145$ , Cys  $\beta 23$ , Cys  $\beta 48$ , Cys  $\beta 106$ , and Ser  $\beta 141$  are coordinated to the P-cluster pair, and these residues are strictly conserved, reflecting their structural importance. Strictly conserved Gly residues (Gly  $\alpha 78$ , Gly  $\alpha 176$ , and Gly  $\beta 47$ ) in the vicinity of the P-cluster pair are also structurally important to avoid steric hindrance with the P-cluster pair. Hydrophilic residues around the P-cluster pair, such as Ser  $\alpha 80$ , Thr  $\alpha 144$ , Glu  $\alpha 175$ , Ser  $\beta 50$ , His  $\beta 52$ , and Gln  $\beta 24$ , are generally not conserved, with the exceptions of Gln  $\beta 46$  and Thr  $\beta 105$ .

#### Mechanism of $H_2$ Evolution

Under optimal conditions, nitrogenase catalyzes the reaction  $N_2 + 8H^+ + 8e^- \rightarrow 2NH_3 + H_2$ .  $H_2$  evolution is thought to arise from  $N_2$  binding and displacing  $H_2$  from the active site (Thorneley & Lowe, 1985). The proportion of electron flux which results in  $H_2$  evolution can increase under conditions where the rate of electron flux decreases, i.e., limiting MgATP

or Fe-protein concentrations (Bishop et al., 1986), or where the pressure of  $N_2$  is low (Hadfield & Bulen, 1969). These findings suggest that there may be an additional  $H_2$  evolution mechanism. Recently, it has been proposed that the disulfide bond in the P-cluster pair may provide a site for  $H_2$  evolution (Chan et al., 1993). Protonation of the doubly reduced P-cluster pair may generate a species that can produce  $H_2$  upon disulfide formation. We propose that there are two  $H_2$  evolution sites in MoFe-protein: the  $N_2$  binding site in the FeMo-cofactor and the disulfide bond site in the P-cluster pair. The aromatic and hydrophobic residues around the P-cluster pair may function in suppressing  $H_2$  evolution activity of nitrogenase by inhibiting proton transfer to the P-cluster pair.

#### *Electron Transfer from the P-Cluster Pair to the FeMo-Cofactor*

The polypeptide environment between the FeMo-cofactor and the P-cluster pair is illustrated in Figure 6b. The edge-edge distance of the FeMo-cofactor to the P-cluster pair is  $\sim 14$  Å, which is consistent with low-resolution studies of the metal center positions in Cp1 (Bolin et al., 1990). Four helices ( $\alpha 54$ – $\alpha 64$ ,  $\alpha 77$ – $\alpha 83$ ,  $\alpha 182$ – $\alpha 196$ , and  $\beta 46$ – $\beta 60$ ) oriented in parallel starting from the P-cluster pair toward the FeMo-cofactor may play an important role in electron transfer. In particular, helices  $\alpha 54$ – $\alpha 64$  and  $\alpha 77$ – $\alpha 83$ , adjacent to the P-cluster pair ligands Cys  $\alpha 53$  and Cys  $\alpha 79$ , provide the most direct polypeptide connection between the P-cluster pair and the FeMo-cofactor. Potential proton-transfer pathways, such as salt bridges and hydrogen-bonding networks, have not been found that could permit the coupling of electron and proton transfer between the P-cluster pair and the FeMo-cofactor. Since the homocitrate is located on the side of the FeMo-cofactor that faces the P-cluster pair, it is possible that the electrons are transferred from the P-cluster pair to the FeMo-cofactor through the homocitrate. Indeed, the importance of homocitrate to the substrate reduction mechanism may arise from its function in the protonation of intermediates and/or participation in the electron-transfer pathway between the P-cluster pair and the FeMo-cofactor, as well as from modulation of the redox properties of the coordinated FeMo-cofactor (Kim & Rees, 1992a).

#### *Fe-Protein Binding Sites*

The  $\alpha$  and  $\beta$  subunits of MoFe-protein are related by an approximate 2-fold axis that passes through the center of the P-cluster pair, and there are two wide and shallow clefts related by this pseudo-2-fold rotation. As the Fe-protein dimer also has a 2-fold NCS axis (Georgiadis et al., 1992), a plausible model for docking the two proteins involves superposition of the Fe-protein's 2-fold axis with the approximate 2-fold axis of MoFe-protein passing through the P-cluster pair (Kim & Rees, 1992b). This docking model, with a stoichiometry of 1:2 MoFe-protein:Fe-protein, is consistent with both mutagenesis and cross-linking results (Govenzensky & Zamir, 1989; Willing & Howard, 1990; Wolle et al., 1992; Thorneley et al., 1993). Three potential contact regions, based on the structures of the MoFe-protein and Fe-protein, include two Fe-protein subunit binding sites (designated A and B in Figure 7a) and a 4Fe:4S cluster binding site (designated C in Figure 7a). Interestingly, the long inserted sequence of Cp1 ( $\alpha 375$ – $\alpha 430$ ) is located around the A-subunit binding site. In

particular, residues  $\alpha 383$ – $\alpha 397$  span the cleft above the FeMo-cofactor, and this region may directly interact with the Fe-protein. Lys  $\alpha 385$ , Asp  $\alpha 387$ , Asp  $\alpha 389$ , and Asn  $\alpha 392$  are exposed to water, and these residues may recognize the *C. pasteurianum* Fe-protein (Cp2) and discriminate against other Fe-proteins such as the *A. vinelandii* Fe-protein (Emerich & Burris, 1978). The location of the long inserted sequences in Cp1 may explain why Cp1 cannot form an active complex with any Fe-proteins other than Cp2 (Emerich & Burris, 1978), and this finding indirectly supports the assignment of Fe-protein binding sites in MoFe-protein and the proposed docking model (Kim & Rees, 1992b).

The potential 4Fe:4S cluster binding site (site C in Figure 7a) and surrounding residues are shown in Figure 7b. Four short helices ( $\alpha 112$ – $\alpha 117$ ,  $\alpha 146$ – $\alpha 150$ ,  $\beta 106$ – $\beta 111$ , and  $\beta 73$ – $\beta 78$ ), which are related by the approximate 2-fold axis, are oriented in parallel from the P-cluster pair toward the protein surface, forming an approximate four-helical bundle upon which the 4Fe:4S cluster of Fe-protein is likely to sit. The edge-edge distance from the P-cluster pair to the end of these helices is about 12 Å; thus the edge-edge distance from the P-cluster pair to the 4Fe:4S cluster of Fe-protein may be about 15 Å. It seems likely that the electron is transferred through these helices. The polypeptide environment around this electron-transfer pathway is primarily provided by aromatic or hydrophobic residues such as Tyr  $\alpha 177$ , Tyr  $\beta 142$ , Leu  $\alpha 149$ , Leu  $\beta 107$ , Ile  $\alpha 150$ , Leu  $\beta 111$ , Phe  $\alpha 117$ , Phe  $\beta 78$ , and Pro  $\alpha 146$ , and they are highly conserved among the known MoFe-protein sequences. There are no unbroken hydrogen-bonding/salt bridge networks along the electron-transfer pathway; therefore, it seems likely that the electron transfer from Fe-protein to the P-cluster pair is not coupled with proton transfer. Polar and charged residues outside the four-helical bundle may be important for interaction with Fe-protein through formation of hydrogen bonds and salt bridges. Asp  $\alpha 153$  and Asp  $\beta 114$ , which are related by the pseudo-2-fold axis, are completely exposed to solvent and are about 18 Å away from the P-cluster pair. His  $\beta 136$  is exposed to solvent and is about 16 Å from the P-cluster pair. Asn  $\alpha 121$ , Glu  $\alpha 112$ , and Glu  $\beta 73$  are also exposed to solvent.

The putative Fe-protein subunit binding sites (sites A and B in Figure 7a), which are related by the pseudo-2-fold rotation, and the exposed residues around these sites are shown in Figures 7c,d. It is intriguing that the FeMo-cofactor is located under one of these possible binding sites. This observation may be relevant to the observation that the Fe-protein is involved in FeMo-cofactor insertion during the MoFe-protein maturation (Robinson et al., 1987). The polypeptide environment around the Fe-protein subunit binding sites is primarily provided by charged or hydrophilic residues located on the MoFe-protein surface. Thus, electrostatic interactions and hydrogen bond formation appear to be important factors in the association of MoFe-protein and Fe-protein, as had been implicated by salt effects on nitrogenase activity (Deits & Howard, 1990).

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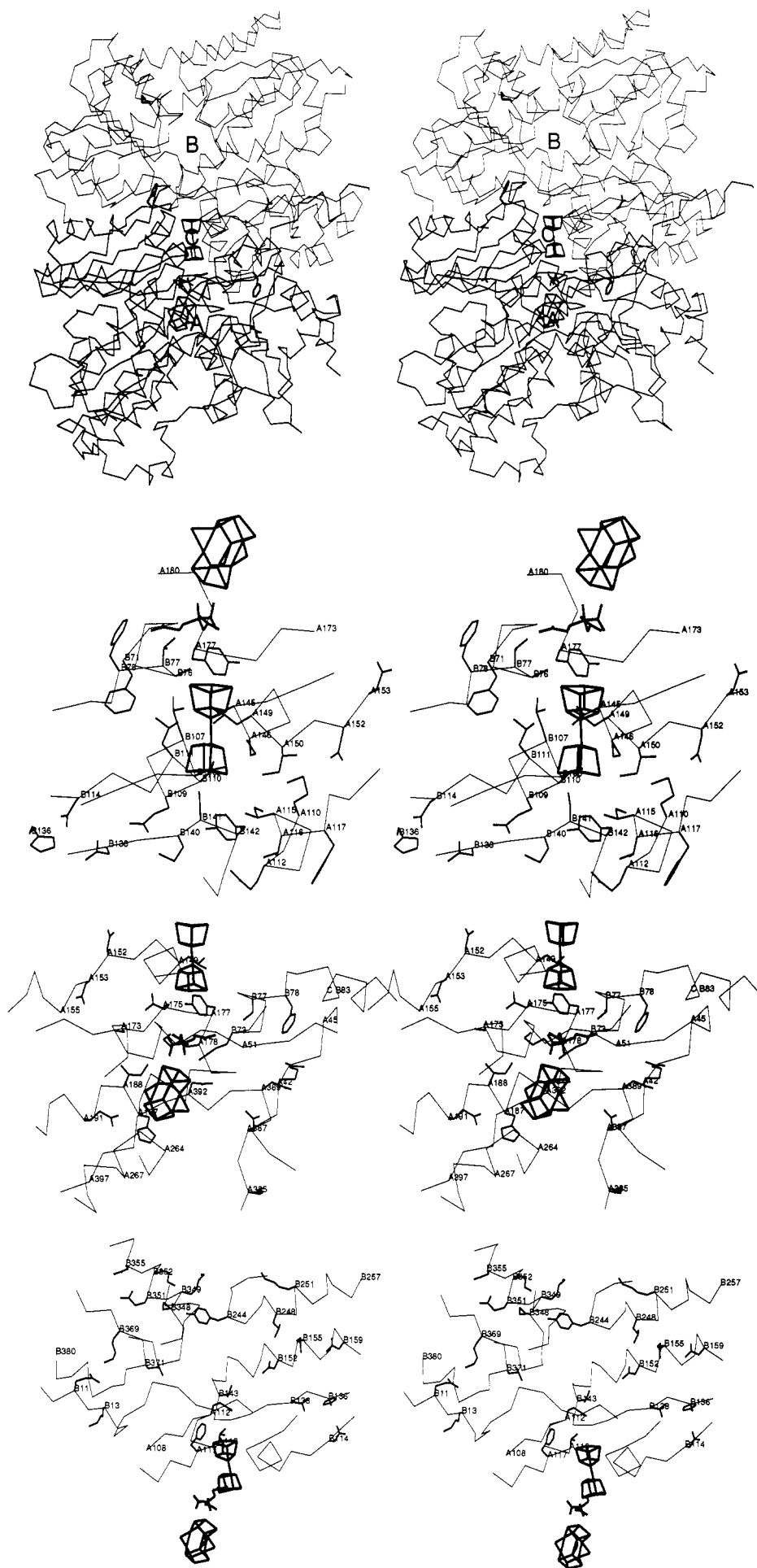




FIGURE 7: From top to bottom: (a) Overview of the Fe-protein binding sites. Three potential contact regions are designated A, B, and C, and side chains of Phe  $\alpha$ 117, Phe  $\beta$ 78, Asp  $\alpha$ 153, Asp  $\beta$ 114, and Met  $\beta$ 348 (Lys  $\beta$ 400 in Av1), which correspond to residues implicated in Fe-protein binding in Av1, are also shown. The  $\alpha$  and  $\beta$  subunits are represented by the thick and thin lines, respectively. (b) Stereoview of the polypeptide environment in the putative Fe-protein cluster binding site (C), near the P-cluster pair. (c) Stereoview of the polypeptide environment in the putative Fe-protein subunit binding site (A) on the  $\alpha$  subunit, near the FeMo-cofactor. (d) Stereoview of the polypeptide environment in the putative Fe-protein subunit binding site (B) on the  $\beta$  subunit.

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